A NEW AND UNEQUIVOCAL SYNTHESIS OF ISOXANTHOPTERIN-6-CARBOXYLIC ACID (CYPRINO-POURPRE B)^{1,2}

Edward C. Taylor* and Riaz F. Abdulla³

Department of Chemistry, Princeton University, Princeton, N.J. 08540 (Received in USA 4 April 1973; received in UK for publication 27 April 1973)

One of the major pteridine constituents of the scales and skin of fresh water fishes, the Cyprinidae, is Cyprino-Pourpre B, which has been identified as isoxanthopterin-6-carboxylic acid (1).⁴ It is believed that 1 arises by biological oxidation of ichthyopterin (2), another of the common pteridine fish-skin pigments.⁵ Although 1 has been synthesized twice previously (by condensation of 2,4,5-triamino-6(1H)-pyrimidinone with diethyl oxomalonate⁶



or with alloxan in alkaline solution⁷), both procedures are inherently ambiguous and, indeed, lead to concomitant formation of varying amounts of the isomeric xanthopterin-7-carboxylic acid. As part of our program directed towards the utilization of pteridine 8-oxides as intermediates for the synthesis of the pteridine natural products,^{8⁻¹⁰} we have achieved a synthesis of isoxanthopterin-6-carboxylic acid (1), totally free of its isomer, by a route which unambiguously positions the substituents in the pyrazine ring.

Thus (see Scheme 1), condensation of aminomalononitrile tosylate with dioximinoacetone gave 2-amino-3-cyano-5-oximinomethylpyrazine 1-oxide (3), which was readily cyclized with guanidine to 2,4-diamino-6-oximinomethylpteridine 8-oxide (4).¹¹ Heating a solution of 4 in a mixture of $POCl_3/DMF$ for 2 hr at 65⁰ resulted in the separation of 5 (93%), which upon 24 hr reflux with 10% aqueous sodium hydroxide solution, followed by acidification, gave isoxanthopterin-6-carboxylic acid (1) (89%), identical both chromatographically and spectroscopically with naturally occurring Cyprino-Pourpre B.



Scheme 1

When 1 was heated at 260° in a current of dry nitrogen, decarboxylation readily occurred (judged complete in 30 min). Dissolution of the dark grey residue in 1 N NaOH followed by decolorization with charcoal and acidification afforded pure isoxanthopterin (6), identical in all respects with an authentic sample.

REFERENCES

- Pteridines. XXXI.; Part XXX: E. C. Taylor and P. A. Jacobi, <u>J.Amer.Chem</u>. <u>Soc</u>., submitted for publication.
- 2. This work was supported in part by a grant (CA-12876) to Princeton University from the National Cancer Institute, National Institutes of Health, Public Health Service.
- 3. Postdoctoral Fellow of the Damon Runyon Memorial Fund for Cancer Research, Inc.
- 4. J. Matsumoto, T. Kajishima, and T. Hama, Genetics, 45, 1177 (1960).
- 5. Y. Mori, J. Matsumoto, and T. Hama, Z. Vergl. Physiol., 43, 531 (1960).
- 6. R. Purrmann, Justus Liebigs Ann. Chem., 548, 284 (1941).
- 7. E. C. Taylor and H. M. Loux, Chem. and Ind., 1585 (1954).
- 8. E. C. Taylor and K. Lenard, J.Amer.Chem.Soc., 90, 2424 (1968).
- 9. E. C. Taylor and K. Lenard, Justus Liebigs Ann. Chem., 726, 100 (1969).
- E. C. Taylor in "The Chemistry and Biology of Pteridines", ed. by K. Iwai, M. Akino, M. Goto, and Y. Iwanami, International Academic Printing Co., Ltd., Tokyo, 1970, pp. 79-93.
- We have previously described the condensation of dioximinoacetone with ethyl α-aminocyanoacetate to give 2-amino-3-ethoxycarbonyl-5-oximinomethylpyrazine 1-oxide, which was a key intermediate in the synthesis of pterin-6-carboxaldehyde (see ref. 9).